

# **Solubility Rules Worksheet With Answers**

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Solubility Rules Worksheet With Answers State whether they are soluble (will dissolve) or insoluble (will not dissolve) in solution. Use solubility rules. Chemical Formula Name Solubility 1.  $\text{NH}_4\text{CH}_3\text{COO}$  Ammonium acetate soluble 2.  $\text{Ba}(\text{OH})_2$  Barium hydroxide soluble 3.  $\text{FeCO}_3$  Iron (II) Carbonate insoluble 4.  $\text{NaOH}$  Sodium hydroxide soluble 5.  $\text{RbNO}_3$  Rubidium nitrate soluble 6. solubility rules worksheet Answers - mrdeakin / FrontPage Answers to Solubility Rules Worksheet. Classify each of the substances as being soluble or insoluble in water. potassium bromide - sol. lead (II) carbonate - insol. barium sulfate - insol. zinc hydroxide - insol. sodium acetate - sol. silver iodide - insol. cadmium (II) sulfide - insol. zinc carbonate - insol. silver acetate - sol USING THE SOLUBILITY RULES dissolve) in solution. Use solubility rules. Chemical Formula Name Solubility 1.  $\text{NH}_4\text{CH}_3\text{COO}$  2.  $\text{Ba}(\text{OH})_2$  3. Iron (II) Carbonate 4.  $\text{NaOH}$  5.  $\text{RbNO}_3$  6. Cesium Sulfate 7.  $\text{MgSO}_4$  8.  $\text{ZnCl}_2$  9. Zinc Hydroxide 10.  $\text{Zn}_3(\text{PO}_4)_2$  11.  $\text{AgBr}$  12.  $\text{KNO}_3$  13.  $\text{Al}_2\text{S}_3$  14. Silver Acetate 15.  $\text{Sr}_2\text{CrO}_4$  16. Aluminum Phosphate 17.  $\text{BaSO}_4$  18.  $\text{Ca}(\text{OH})_2$  19.  $\text{BaCO}_3$  20.  $\text{MgCrO}_4$  Chemical Formula Name Solubility - Laney College Solubility Rules Worksheet Period: \_\_\_\_\_ 1. Classify each of the substances as being soluble or insoluble in water. a.  $\text{KBr}$  = b.  $\text{PbCO}_3$  = c. zinc hydroxide = d. sodium acetate = e. silver iodide = f. zinc carbonate = g. silver acetate = h. copper (II) sulfide = i.  $\text{Mg}_3(\text{PO}_4)_2$  = j.  $\text{KOH}$  = k.  $\text{NH}_4\text{OH}$  = l.  $\text{Hg}_2\text{SO}_4$  = m.  $\text{PbI}_2$  = . 2. Solubility Table worksheet - Weebly Solubility Rules. \* Description/Instructions. There are rules

that determine whether a compound is soluble or not. They are as follows: 1. All common salts of the Group 1 elements and the ammonium ion are soluble. 2. All common acetates and nitrates are soluble. 3. Solubility Rules Quiz - Free Math worksheets, Free phonics ... Solubility And Answers. Displaying top 8 worksheets found for - Solubility And Answers. Some of the worksheets for this concept are Name sec date chem 1319 ws16 solubility work, Solubility work 2 level 1, Solubility product work, Work solubility graphs name, Work 7more solubility problems answer key, Solubility rules work, Use the provided solubility graph to answer the following, Solubility ... Solubility And Answers Worksheets - Learny Kids Simple Solubility Rules: -Nitrate ( $\text{NO}_3^-$ ) salts are soluble. -Alkali (group 1A) salts and  $\text{NH}_4^+$  are soluble. -Cl-, Br-, and I-salts are soluble (NOT  $\text{Ag}^+$ ,  $\text{Pb}^{2+}$ ,  $\text{Hg}_2^{2+}$ ) -Sulfate salts are soluble (NOT  $\text{BaSO}_4$ ,  $\text{PbSO}_4$ ,  $\text{HgSO}_4$ ,  $\text{CaSO}_4$ ) -OH- salts are only slightly soluble (NaOH, KOH are soluble,  $\text{Ba}(\text{OH})_2$ ,  $\text{Ca}(\text{OH})_2$  are marginally soluble) - $\text{S}^{2-}$ ,  $\text{CO}_3^{2-}$ ,  $\text{CrO}_4$ ,  $\text{PO}_4$  Solubility Rules Made Easy SOLUBILITY RULES 1. Salts of ammonium ( $\text{NH}_4^+$ ) and Group IA are always soluble. 2. Solubility Rules - Michigan State University Answer Key Soluble. Displaying top 8 worksheets found for - Answer Key Soluble. Some of the worksheets for this concept are Work 7more solubility problems answer key, Use the provided solubility graph to answer the following, Name sec date chem 1319 ws16 solubility work, Solubility rules work, Abcs of vitamins, Solubility graph work, Physical and chemical changes work, 10 1314 more on ... Answer Key Soluble Worksheets - Learny Kids In this science worksheet, your child learns about

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1) What is the concentration of a saturated silver (I) acetate solution?  $K_{sp}(\text{AgC}_2\text{H}_3\text{O}_2) = 1.94 \times 10^{-3}$ . Since  $K_{sp} = [\text{Ag}^+][\text{C}_2\text{H}_3\text{O}_2^-]$ , and the concentration of silver ions is the same as the concentration of acetate ions, we can set up the following equation:  $1.94 \times 10^{-3} = x^2$   $x = 0.0440$  M Solubility Product

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sec date chem 1319 ws16 solubility work, Solubility rules work table a. Solubility Rules Worksheets - Kiddy Math Worksheet #5 ~ Solubility Rules IO Instructions: Use the solubility rules from below to put an "X" in the boxes that are SOLUBLE. Solubility Rules: 1. All common salts of Group 1 elements and ammonium are soluble. 2. All common acetates and nitrates are soluble. 3. Solubility Rules Worksheet KEY - studylib.net worksheet : For each of the following combinations of reactants, do the Precipitation Reactions Solubility Rules Lab Answers Using the Precipitation Reactions and Solubility Rules Chemistry Laboratory Kit, students perform chemical reactions by combining sets of salt solutions, generate lists [DOC] Chemistry Precipitation Reactions Answer Key Soluble And Insoluble Salts. Soluble And Insoluble Salts - Displaying top 8 worksheets found for this concept.. Some of the worksheets for this concept are Solubility rules name chem work 15 1, Name sec date chem 1319 ws16 solubility work, How soluble is it, Solubility rules work, Solubility, Net ionic equations name advanced chem work 10 4, Solubility rules, Electrolytes work name key.

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