

# Chapter Seventeen Electrochemistry Cengage

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## Chapter Seventeen Electrochemistry

### Cengage CHAPTER SEVENTEEN ELECTROCHEMISTRY

For Review 1. Electrochemistry is the study of the interchange of chemical and electrical energy. A redox (oxidation-reduction) reaction is a reaction in which one or more electrons are transferred. In a galvanic cell, a spontaneous redox reaction occurs which produces an electric current. In an CHAPTER

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chapter. Chapter 17 covers the following topics: 17.1

Galvanic Cells. Cell Potential 17.2 Standard Reduction

Potentials. Line Notation Complete Description of a

Galvanic Cell 17.3 Cell Potential, Electrical Work, and

Free Energy Chapter 17 - Electrochemistry -

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Electrolytic cells 5. Commercial cells Electrochemistry • Electrochemistry is the study of the conversion of electrical and chemical energy • The conversion takes place in an electrochemical Chapter 17 [□□□□] Section

18.1 Balancing Oxidation-Reduction Equations

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Oxidation ... Chapter 18 Electrochemistry -

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Electrochemistry: Animations and Documents:

Electrochemical Cell Part 1 This animation shows the components of a zinc-copper cell. Electrochemical Cell

Part 2 This animation shows the lost of electrons in the zinc half-cell. Page One - CORE\_Chapter Seventeen -

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